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4.2 The Nature Of Aqueous Solutions: Strong And Weak Electrolytes • Classify Many Substances As Strong, Weak Or Nonelectrolytes. • Solute: 1. If It And The Solvent Are Present In The Same Phase, It Is The One In Lesser Amount. 2. Feb 13th, 2024.

Tutorial 4 SOLUTION STOICHIOMETRY  
20) Acidic; 0.0758 M K<sub>3</sub>PO<sub>4</sub> 5) 20.8 ML AgNO<sub>3</sub> 21) 247 ML KOH 6) 166 ML AlCl<sub>3</sub> 22) 0.0556 M BaCl<sub>2</sub> 7) Mar 6th, 2024  
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Stoichiometry – Cont. 8 Example 4.8: What Volume Of 0.150 M KCl Is Required To Completely React With 0.150 L Of 0.175 M  $\text{Pb}(\text{NO}_3)_2$  In The Reaction  $2 \text{KCl}(\text{aq}) + \text{Pb}(\text{NO}_3)_2(\text{aq}) \rightarrow \text{PbCl}_2(\text{s}) + 2 \text{KNO}_3(\text{aq})$ ? 0.1 May 7th, 2024 Chapter 4. Aqueous

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