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Homework: Solubility And Complex Ion Equilibria 1.) The Molar Solubility Of PbI2 Is  $1.51 \times 10$ –3 M. Calculate The Value Of K Sp For PbI2. ( $1.38 \times 10$ –8) 2.) It Is Observed That 7.53 Mmol O Jan 9th, 2024CHEMISTRY 12 SOLUBILITY REVIEW QUESTIONSAnswers To Solubility Problem Set 2 (for Questions On The Other Side):  $1.3.7 \times 10$ –19 2. [Ca2+] = [SO 4 2-] =  $8.4 \times 10$ -3 M 3. [Ca2+] =  $2.3 \times 10$ -4 M, [F-] =  $4.6 \times 10$ -4 4.  $8.5 \times 10$ -15 G/L 5. K TIP =  $5.8 \times 10$ -7 Ppt Forms 6. K TIP =  $9.4 \times 10$ -7 Ppt Forms 7. A) 2.5 X 10-5 B) 3.7 X 10-9 8. K Sp (at Temperature X OC) =  $1.1 \times 10$ -8, X OC >  $25 \times 10$ -3 M (pure Water),  $1.5 \times 10$ -4 M ( $0.10 \times 10$  M Pb(NO ... Feb 15th, 2024. SOLUBILITY CURVES - PTHS HONORS CHEMISTRYSolubility Curves Study The Solubility Curves In The Figure, And Then Answer The Questions That Follow. 1. What Relationship Exists Between Solubility And Temperature For Most Of The Substances Shown? 2. A. What Is The Exception? NN T I B. What General Principle

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Accounts For This Ex-ception? 6 6" 3. A. Approximately How Many Grams Of NaNO3

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