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Th Reading For Lecture #21 II. I. SOLUTIONS AND SOLUBILITY ...
3. Lewis Acid And Base - More General Definition - Applies To Reactions That Don't Involve A Hydrogen Ion . Lewis . Base -Lewis . Acid -Example 1 . $\text{F}_3\text{B} + \text{NH}_3 \rightarrow \text{F}_3\text{B} \cdot \text{NH}_3$. Ammonia Is The Lewis Base. It Donates Lone-pair Electrons To BF_3 , The Lewis Acid And Jan 15th, 2024.
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