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2. Q. List Three Subatomic Particles Of An Atom. Compare Them On The Basis Of Relative Mass And Charge In A Tabular Form. Ans: Particle Symbol Charge Mass Electron  $e^-$   $1.60 \times 10^{-19}$  C  $9.1 \times 10^{-31}$  Kg Proton  $+p$  ( $\text{H}^+$ )  $1.60 \times 10^{-1}$  Feb 1th, 2024

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(c) Three Mole (d) Four Moles (4) Which Of The Following Weighs The Most? (a) One G-atom Of Nitrogen (b) One Mole Of Water (c) One Mole Of Sodium (d) One Molecule Of  $\text{H}_2\text{SO}_4$  (5) 5.0 Litre Of Feb 2th, 2024

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SAMPLE - SAMPLE - SAMPLE - SAMPLE SAMPLE - SAMPLE - SAMPLE - SAMPLE SAMPLE - SAMPLE - SAMPLE - SAMPLE Nationality - Ex: American/USA Your Birthday Country Of Birth If You Had Other Citizenship At Birth Day, Month, Year City & State First And Middle Name This Is A SAMPLE Application. Your D Feb 2th, 2024

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Question No: 13 ( Marks: 1 ) -Please Choose One On Balance Volume Technique Of Technical Analysis Was Developed By: Charles Dow Joseph Granville John Bollinger Welles Wilder Joseph Granville Created An On-balance Volume Technique In 1976, In His Classic "New Strategy Of Daily Mar 1th, 2024

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