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Solvent, A Solution Is Formed If The Solvent Is Water, An Aqueous Solution Is Formed The Concentration Of A Solution (mol dm<sup>-3</sup>) Shows The Amount Of Solute

Dissolved In A Given Volume Of Solution Standard Solution: Solution Whose Concentration Is Accurately Known  $[X] = \frac{\text{Amt Of X (mol)}}{V \text{ Of lth}}$ , 2024

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Directions: Solve Each Of The Following Problems.

Show Your Work, Including Proper Units, To Earn Full

Credit. 1. Silver And Nitric Acid React According To The Following Balanced Equation: 3 A 9th, 2024

### **Stoichiometry: Problem Sheet 2 -**

#### **Teachnlearnchem.com**

Chemistry: Stoichiometry - Problem Sheet 2 KEY 9) 2

24 2 2 23 2 2 2 2 4.63 X 10molecules | 1 Mol | 6.02 X

10 Molecules| 1 Mol Cl 1mol 71 G Cl Cl X 546 G Cl 10)

292 G Ag 1 Mol Ag 108 G Ag 1 Mol Cu 1 Mol Ag 63.5 G

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#### **Answers**

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Chemistry: Stoichiometry - Practice Problem Sheet 2

Directions: Solve Each Of The Following Problems.

Show Your Work, Including Proper Units 1.  $\underline{\quad}$  CaCl<sub>2</sub> +

$\underline{\quad}$  AgNO<sub>3</sub>  $\underline{\quad}$  Ca(NO<sub>3</sub>)<sub>2</sub> +  $\underline{\quad}$  AgCl How Many

Grams Of Silver Chloride Are Produced When 45 G Of Calcium Chloride React With Excess Silver ... 3th, 2024

### **Chemistry Stoichiometry Problem Sheet 2**

#### **Answer Key**

Stoichiometry: Problem Sheet 1 - FREE Chemistry

Materials Chemistry: Stoichiometry - Problem Sheet 1

Directions: Solve Each Of The Following Problems.

Show Your Work, Including Proper Units, To Earn Full

Credit. 1. Silver And Nitric Acid React According To The Following Balanced Equation:  $3 \text{ Ag(s)} + 4 \text{ HNO}_3 \text{ (aq)} \rightarrow 3 \text{ AgNO}_3 \text{ (aq)} + 2 \text{ H}_2 \text{ O(l)} + \text{ NO ...}$  9th, 2024

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