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SO₃ As A Conversion Factor, We Determine The Mass That This Number Of Moles Of SO₃ Has. The First Step Resembles The

Exercises We Did In Section 6.4 "Mole-Mole Relationships In Chemical Reactions". As Usual, We Start With The Quantity We Were Given: $3.59 \text{ Mol Fe}_2\text{O}_3 \times 3 \text{ Mol SO}_3 / 1 \text{ Mol Fe}_2\text{O}_3 = 10.77 \text{ Mol SO}_3$ The Mol Fe₂O₃ Units Cancel, Leaving Mol SO₃ ... Jun 1th, 2024

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