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4. Draw The Full Structure, Given The Condensed Structure. (Note: CH 3CH 2OH (CH 3) 2CH 2NH 2 CH 2CHCl CH 3CHO CH 3CO 2H 5. Fill In The Full Structure, Including Attached Hydrogens And Attached Lone Pairs, For The Following Line-angle Structures. If Given A Condensed Structure, Convert It To A Line-angle. OH O CH 3CH 2CH 2CH 3 CH 3CO 2H Jan 3th, 2024

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Cotx = $-\csc 2 X$ (11) D Dx Secx = Secxtanx (12) D Dx Cscx = $-\csc x$ (13) D Dx Ex = Ex (14) D Dx Ax = Ax Lna (15) D Dx Ln|x| = 1 X (16) D Dx Sin-1 X = 1 $\sqrt{1-x^2}$ (17) D Dx Cos-1 X = $-1 \sqrt{1-x^2}$ (18) D Dx Tan-1 X = 1 X2 +1 (19) D Dx Cot-1 X = -1 X2 +1 (20) D Jan 1th, 2024

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Substituents: The Carbon Branches From The Main Chain (root) Guidelines For Naming Branched-Chain Hydrocarbons. Rule 1: Find The Longest Continuous C -chain And Assign The Root Name. Rule 2: Assign The Suffix Based On The Type Of Carbon -carbon Bonds. Rule 3: Name Each Substitu Feb 3th, 2024

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Draw The Correct Structural Formulas

For Example Ch 3 Or Ch 2 Instead Of Showing All These Bonds, Draw A Correct Lewis Structure For Ch3 2chcooh See 35 The Two Ch3 S Go At The Ends Vertically In The Middle Is The Linear Hcoh With A Double Bonded O At The Second Carbon 36 Draw Jun 1th, 2024

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2.8 Formulas, Names And Masses Structural Formula Of ...

HCl As Ionic Or Covalent And Identify The Ions If Present. Problem: • Name The Compounds: PF5, CrF3, N2O3, Fe2O3 – Molecular →phosphorus Pentafluoride – Ionic, F-, Cr3+ →chromium(III) Fluoride – Molecular →dinitrogen Trioxide – Ionic, O2-, Fe3+ →iron(III Jan 2th, 2024

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