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Thermochemistry, Thermodynamics

Thermochemistry

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Chemical Potential Energy Is Energy Stored In
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$\text{C}_2\text{H}_5\text{OH}(l) + 3 \text{ O}_2(g) \rightarrow 2 \text{ CO}_2(g) + 3 \text{ H}_2\text{O}(g)$ $\Delta H^\circ_{\text{rxn}} = -1367 \text{ kJ/mol}$
A 100.0 mL Sample Of 0.300 M NaOH Is Mixed With A 100.0 mL
Sample Of 0.305 M HNO_3 In A Coffee Cup
Calorimeter. If Both Solutions Were Initially At 35.0 °C
And The Temperature Of The Resulting Solution Was
Recorded As 37.0 °C, Determine The ΔH_{rxn} (in Units Of
kJ/mol). 3th, 2024

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Name_____ MULTIPLE CHOICE. Choose The One
Alternative That Best Completes The Statement Or
Answers The Question. 1) A Chemical Reaction That
Absorbs Heat From The Surroundings Is Said To Be
_____ And Has A _____ ΔH At Constant Pressure.
A)endothermic, Positive 3th, 2024

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Name: _____ Section: _____ Chapter 8 Worksheet Spring 2007 Page 1 Of 4 Chapter 10 Practice Worksheet:

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