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$C_2H_5OH(l) + 3 O_2(g) \rightarrow 2 CO_2(g) + 3 H_2O(g)$  Calor 31) A 100.0 mL  
Sample Of 0.300 M NaOH Is Mixed With A 100.0 mL  
Sample Of 0.305 M  $HNO_3$  31) In A Coffee Cup  
Calorimeter. If Both Solutions Were Initially At 35.0 °C  
And The Temperature Of The Resulting Solution Was  
Recorded As 37.0 °C, Determine The  $\Delta H_{rxn}$  (in Units Of  
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A) endothermic, Positive Mar 1th, 2024

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