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1 Chemical Reactions 1.1 Rate Of Reaction And Dependence On Temperature We Will Once Again Look At The Formation Of Ammonia ( $\text{NH}_3$ ) From Nitrogen And Hydrogen (see Section Chemical Equilibrium Of The Thermodynamics Chapter). This Reaction Follows The Equation:  $\text{N}_2 + 3\text{H}_2 \rightleftharpoons 2\text{NH}_3$  (1)  $H^\circ = 92 \text{ KJ Mol}^{-1}$   $S^\circ = 192 \text{ J Mol}^{-1} \text{ K}^{-1}$  May 2th, 2024

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And Similarity. Principles Of Inviscid And Real Fluid  
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