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4. $\text{HCl(aq)} + \text{Sr(OH)}_2\text{(aq)} \rightarrow \text{SrCl}_2\text{(aq)} + \text{H}_2\text{O(l)}$ 5. $\text{AlBr}_3\text{(s)} \rightarrow \text{Al(s)} + \text{Br}_2\text{(l)}$
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Polysaccharide Starch, Cellulose, Glycogen (animal) Benedicts-glucose Iodine-starch Amino Acid Protein, Polypeptide Hair, Enzyme, Hemoglobin, Insulin Ninhydrin, Biuret Nucleotide Nucleic Acid DNA, RNA Methyl Green Fatty Acid/glycerol Fat Or Tri - Glyceride Cooking Oil, Butter Grease Test With Brown Paper 2th, 2024

Chemical Reactions Describing Chemical Reactions

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Chemical Equations Types Of Chemical Reactions

Balancing Chemical Equations Process For Balancing Chemical Equations: 1. Determine Correct Chemical Formulas Of All Reactants And Products 2. Start With "heavier" Atoms—balance Number Of These On Reactant And Product Sides Of Equation 3. If Elements Appear In Equati 2th, 2024

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Over A Bunsen Burner Strongly For About 3 Minutes. Insert A Burning Wood Splint Into The Test Tube. If Carbon Dioxide Gas (CO₂) Is Present, It Will Put The Flame Out. Note Any Change In The Appearance Of The Residue In The Test Tube. DISPOSAL: This May Safely Go Down The D 2th, 2024

LAB: TYPES OF CHEMICAL REACTIONS

E. Turn On The Methane Gas And Light The Bunsen Burner. Open The Air Holes To Add More Oxygen And Make A Blue Cone. (Use This Flame For The Next Trial.) Observation Word Equation Balanced Formula Equation Type Of Reaction F. Place A Small 3th, 2024

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