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(PO 4)  $2 + 6\text{H}_2\text{O}$  Double Replacement 7. Copper And Sulfuric Acid React To Form Copper (II) Sulfate And Water And Sulfur Dioxide  $\text{Cu} + 2\text{H}_2\text{SO}_4 \rightarrow \text{CuSO}_4 + 2\text{H}_2\text{O} + \text{SO}_2$  Single Replacement And Acid Base 8. Mar 3th, 2024.

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Balance The Equations Below: 1)  $1\text{N}_2 + 3\text{H}_2 \rightarrow 2\text{NH}_3$  2)  $2\text{KClO}_3 \rightarrow 2\text{KCl} + 3\text{O}_2$  3)  $2\text{NaCl} + 1\text{F}_2 \rightarrow 2\text{NaF} + 1\text{Cl}_2$  4)  $2\text{H}_2 + 1\text{O}_2 \rightarrow 2\text{H}_2\text{O}$  5)  $1\text{Pb(OH)}_2 + 2\text{HCl} \rightarrow 2\text{H}_2\text{O} + 1\text{Pb}$

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